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## Li-Ping Deng and Yong-Zhou Hu\*

Department of Medicinal Chemistry, College of Pharmaceutical Science, Zhejiang University, Hangzhou 310031, Zhejiang, People's Republic of China

Correspondence e-mail: huyz@zjuem.zju.edu.cn

#### **Key indicators**

Single-crystal X-ray study T = 295 K Mean  $\sigma$ (C–C) = 0.004 Å R factor = 0.055 wR factor = 0.152 Data-to-parameter ratio = 16.6

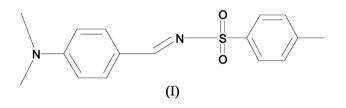
For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e.

# *N*-[4-(Dimethylamino)benzylidene]-4-methylbenzene-sulfonamide

The molecule of the title compound,  $C_{16}H_{18}N_2O_2S$ , consists of essentially planar 4-(dimethylamino)benzylidene and 4-tolyl fragments bonded through a sulfone S atom, which is approximately coplanar with both fragments. The mean planes of the MeC<sub>6</sub>H<sub>4</sub>S and Me<sub>2</sub>NC<sub>6</sub>H<sub>4</sub>CH=NS groups are roughly orthogonal and form a dihedral angle of 101.28 (9)°.

#### Comment

*N*-Sulfonylaldimines attract the attention of organic chemists because of their use as synthetic reagents (Love *et al.*, 1994). As electron-deficient imines, they find elegant application in inverse electron demand Diels–Alder chemistry (Boger *et al.*, 1991), as well as enophiles in stereochemically controlled ene reactions (Melnick *et al.*, 1988). Sulfonylaldimines have also been shown to possess thrombin inhibitor activity (Supuran *et al.*, 2000). Our interest in sulfonylaldimine derivatives is driven both by their biological and by their synthetic capabilities. The title compound, (I), has been prepared and studied in order to obtain a better understanding of its reactivity.



The molecular structure of the title compound is shown in Fig. 1. All non-H atoms of its molecule, with the exception of sulfone atoms O1 and O2, belong to one of the two almost planar fragments,  $MeC_6H_4S$  or  $Me_2NC_6H_4CH$ —NS, which share the sulfone S1 atom. The mean planes of the two groups are roughly orthogonal; they form a dihedral angle of 78.72 (9)°.

### **Experimental**

4-Dimethylaminobenzaldehyde oxime (1 mmol) was refluxed in ethanol (20 ml) with sodium chloro(*p*-tosyl)amide (chloramine-T; 1.2 mmol) for 4 h, and the reaction mixture was then cooled to room temperature. After removal of the solvent, a yellow solid product was obtained, which was washed with water (30 ml) and extracted with dichloromethane (30 ml). The extracts were dried over anhydrous sodium sulfate and concentrated in a vacuum, and the residue was recrystallized from ethanol to give the title compound. Diffraction quality crystals were obtained by slow evaporation of an acetone/ ethanol solution at room temperature.

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#### Crystal data

 $C_{16}H_{18}N_2O_2S$   $M_r = 302.39$ Monoclinic,  $P2_1/c$  a = 17.1543 (4) Å b = 8.3835 (2) Å c = 10.9679 (3) Å  $\beta = 105.976 (1)^{\circ}$   $V = 1516.41 (6) \text{ Å}^3$  Z = 4

#### Data collection

Rigaku R-AXIS RAPID
diffractometer
$\omega$ scans
Absorption correction: multi-scan
(ABSCOR; Higashi, 1995)
$T_{\min} = 0.911, T_{\max} = 0.957$
9646 measured reflections

#### Refinement

Refinement on $F^2$ $R[F^2 > 2\sigma(F^2)] = 0.055$ $wR(F^2) = 0.152$ S = 1.00	$\begin{split} w &= 1/[0.0012F_{o}^{2} + 1.0\sigma(F_{o}^{2})]/(4F_{o}^{2}) \\ (\Delta/\sigma)_{max} &< 0.001 \\ \Delta\rho_{max} &= 0.48 \text{ e } \text{\AA}^{-3} \\ \Delta\rho_{min} &= -0.39 \text{ e } \text{\AA}^{-3} \end{split}$
3176 reflections 191 parameters	Extinction correction: Larson (1970), equation 22
H-atom parameters constrained	Extinction coefficient: 88 (2)

 $D_x = 1.324 \text{ Mg m}^{-3}$ 

Cell parameters from 9315

 $0.30 \times 0.22 \times 0.20$  mm

3458 independent reflections 1718 reflections with  $I > 2\sigma(I)$ 

Mo  $K\alpha$  radiation

reflections

 $\theta = 1.2-27.5^{\circ}$  $\mu = 0.22 \text{ mm}^{-1}$ 

T = 295 (1) K

Prism, yellow

 $\begin{aligned} R_{\rm int} &= 0.054 \\ \theta_{\rm max} &= 27.5^\circ \end{aligned}$ 

 $h = -20 \rightarrow 22$ 

 $k = -10 \rightarrow 10$ 

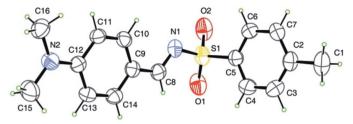
 $l = -14 \rightarrow 13$ 

## Table 1

Selected geometric parameters (Å, °).

S1-O1	1.435 (2)	N1-C8	1.289 (4)
S1-O2	1.426 (2)	N2-C12	1.355 (4)
S1-N1	1.645 (2)	N2-C15	1.447 (4)
S1-C5	1.755 (2)	N2-C16	1.435 (3)
O2-S1-O1	118.6 (1)	C5-S1-O2	109.0 (1)
N1-S1-O1	112.2 (1)	C5-S1-N1	101.6 (1)
C5-S1-O1	108.4 (1)	S1-N1-C8	117.3 (2)
N1-S1-O2	105.6 (1)	S1-C5-C4	119.9 (2)
O1-S1-N1-C8	17.4 (2)	O1-S1-C5-C6	164.4 (2)
O2-S1-N1-C8	148.0 (2)	O2-S1-C5-C4	-149.5(2)
C5-S1-N1-C8	-98.2(2)	N1-S1-C5-C6	-77.2 (2)
O1-S1-C5-C4	-19.1(3)	S1-N1-C8-C9	179.8 (2)

All H atoms were positioned geometrically. The methyl H atoms were then constrained to an ideal geometry, with C-H distances of



#### Figure 1

View of the molecule of the title compound showing the atom-labelling scheme. Displacement ellipsoids are drawn at the 30% probability level.

0.96 Å and  $U_{\rm iso}({\rm H}) = 1.2U_{\rm eq}({\rm C})$ , but each group was allowed to rotate freely about its C–C bond. All other H atoms were placed in geometrically idealized positions and constrained to ride on their parent atoms with C–H distances of 0.98 Å and  $U_{\rm iso}({\rm H}) = 1.2U_{\rm eq}({\rm C})$ . For the refinement, only data with  $\theta > 27^{\circ}$  was used.

Data collection: *PROCESS-AUTO* (Rigaku Corporation, 1998); cell refinement: *PROCESS-AUTO*; data reduction: *CrystalStructure* (Rigaku/MSC, 2004); program(s) used to solve structure: *SIR97* (Altomare *et al.*, 1999); program(s) used to refine structure: *CRYS-TALS* (Betteridge *et al.*, 2003); molecular graphics: *ORTEP-3 for Windows* (Farrugia, 1997); software used to prepare material for publication: *CrystalStructure*.

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